Palladium-Catalyzed Disilylation of *o*-Quinodimethanes: Synthesis of 9- and 10-Membered Disilacarbocycles

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ABSTRACT



o-Quinodimethanes are efficiently inserted into a silicon-silicon bond of cyclic disilanes in the presence of a palladium-diphenyl-2pyridylphosphine catalyst, giving 9- and 10-membered disilacarbocycles, that is, benzodisilonine and benzodisilecine.

Much attention has been directed to synthetic utilization of o-quinodimethanes as an efficient 4-carbon unit in constructing 6-membered carbocyclic frameworks via [4 + 2] cycloaddition (Diels–Alder reaction).¹ Although the wide applicability of the cycloaddition enables diverse functional molecules including steroids,² alkaloids,³ polyacenes,⁴ and fullerenes⁵ to be synthesized in a straightforward manner, little has been known about a different type of transformation using *o*-quinodimethanes,^{6,7} which would lead to production of cyclic structures of other sizes, despite the significant synthetic potential arising from their highly reactive character.

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We have recently reported on the palladium-catalyzed distannylation of *o*-quinodimethanes, which provides α, α' -distannyl-*o*-xylenes of structural diversity, exemplifying that the transient carbon–carbon double bond can undergo a facile transition-metal-catalyzed insertion reaction into an element–element σ -bond.⁸ In this context, we have been investigating the synthetic utility of this protocol and have found that a silicon–silicon bond of disilanes is likewise added across *o*-quinodimethanes smoothly. Herein we disclose that *o*-quinodimethanes serve as an efficacious four-

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Table 1. Palladium-Catalyzed Disilylation of o-Quinodimethanes with $2a^{a}$



^{*a*} The reaction was carried out in dioxane (1 mL) at 40 °C using **1** (0.40 mmol), **2a** (0.20 mmol), KF (0.74 mol), and 18-crown-6 (0.74 mmol) in the presence of Pd(dba)₂ (0.010 mmol) and Ph₂P(2-Py) (0.010 mmol). ^{*b*} Isolated yield based on **2a**. ^{*c*} 60 °C.

carbon unit in constructing 9- and 10-membered disilacarbocycles, that is, benzodisilonine and benzodisilecine, through palladium-catalyzed disilylation of 5- and 6-membered cyclic disilanes.^{9,10}

Disilylation of o-quinodimethanes, generated in situ from 2-[(trimethylsilyl)methyl]benzyl phenyl carbonates (1) and a fluoride ion,^{11,12} with naphthalene-fused disilacyclopentane **2a** could be achieved by the use of a catalytic amount of bis(dibenzylideneacetone)palladium diphenyl-2-pyridylphosphine complex, which is also effective for the above distannylation. As depicted in Table 1, a 9-membered disilacarbocycle, benzodisilonine **3aa**, was formed straightforwardly via insertion of an *exo*-1,3-diene moiety of simple *o*-quinodimethane (from **1a**) into the silicon—silicon bond of **2a** (entry 1). Such substituted *o*-quinodimethanes as 3-phenyl- (from **1b**), 4-fluoro- (from **1c**), or 4-chloro-*o*-quinodimethane (from **1d**) took part in the disilylation efficiently to afford the respective products



(3ba-3da) in 64%, 59%, or 57% yield (entries 2–4), whereas the reaction using methyl-substituted *o*-quino-dimethanes (from 1e-1g) became somewhat sluggish (entries 5–7).

The present insertion reaction was also applicable to disilacyclopentane **2b**, giving moderate yields of a benzodisilonine (**3ab** or **3bb**) (Scheme 1). Furthermore, a 10membered disilacarbocycle, benzodisilecine (**3ac** or **3bc**), could be synthesized by treatment of biphenyl-fused disilacyclohexane **2c** with *o*-quinodimethanes, while the reaction of disilacyclohexane **2d** furnished **3ad** in only 8% yield. Although the yield was slightly lower than those observed with cyclic disilanes, an acyclic disilane (**2e**) participated in the reaction to provide a disilyl-*o*-xylene (**3ae**).

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Scheme 2. Disilylation via Cross-Coupling



Because a benzylic carbonate moiety in **1** should be prone to undergo a cross-coupling reaction with **2** in the presence of a fluoride ion and a palladium catalyst,^{13,14} an *o*-quinodimethane intermediate might not be involved in the present disilylation: cross-coupling of **1** at a C $-OCO_2Ph$ moiety with **2** followed by fluoride-ion-induced siliconsilicon exchange between the resulting 2-[(trimethylsilyl)methyl]benzylsilane and SiOCO₂Ph (and/or SiOPh) (Scheme 2). However, this pathway can be excluded since the reaction using **1h** or **1i**, a regioisomer of **1a**, did not afford the respective products (Scheme 3). In addition,



Scheme 4 describes the indispensability of the palladium catalysis for the disilylation. Thus, disilylation product **3aa** was not formed at all when the reaction of **1a** with **2a** was

Scheme 4.	Reaction	of 1a	with	2a	in	the	Absence	of	the
Palladium Catalyst									







carried out in the absence of the palladium catalyst. This result also rules out a pathway where a fluoride ion acts as a catalyst, as previously reported in the disilylation of 1,3-dienes.¹⁵

These results prompt us to propose a catalytic cycle that is triggered by formation of palladacycle 4^{16} from an *o*-quinodimethane and a Pd(0) complex (Scheme 5, Cycle A). The resulting palladacycle then reacts with 2 to produce benzylpalladium complex 5, which undergoes reductive elimination of 3 with regenerating the Pd(0) complex.¹⁷ Although metal-catalyzed disilylation of such unsaturated carbon-carbon compounds as alkynes and alkenes generally involves oxidative addition of 2 to a Pd(0) complex as a key step,^{10b} we are now in position that Cycle A would be more plausible than Cycle B, since a stoichiometric reaction of 2a with the Pd(0)-Ph₂P(2-Py) complex did not give oxidative adduct 6 (Scheme 6).

In conclusion, disilylation of *o*-quinodimethane has been accomplished by use of a palladium–Ph₂P(2-Py) complex,

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and various 9- and 10-membered disilacarbocycles, which would be otherwise difficult to be synthesized, are accessible directly. Further studies on catalytic insertion reactions of o-quinodimethanes into other element—element σ -bonds are in progress in our laboratory.

Supporting Information Available: Experimental procedure and characterization of products. This material is available free of charge via the Internet at http://pubs.acs.org.

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